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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/698,607	10/30/2003	Sivapackia Ganapathiappan	10010060-5	9838
7590 03/02/2005		EXAMINER		
HEWLETT-PACKARD COMPANY			ZALUKAEVA, TATYANA	
Intellectual Property Administration P.O. Box 272400 Fort Collins, CO 80527-2400			ART UNIT PAPER I	
			1713	

DATE MAILED: 03/02/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

Office Action Summary		Application No.	Applicant(s)			
		10/698,607	GANAPATHIAPPAN, SIVAPACKIA			
		Examiner	Art Unit			
		Tatyana Zalukaeva	1713			
Period for I	The MAILING DATE of this communication app Reply	ears on the cover sheet with the c	orrespondence address			
THE MA - Extension after SIX - If the period of the period	RTENED STATUTORY PERIOD FOR REPLY ALLING DATE OF THIS COMMUNICATION.  Ins of time may be available under the provisions of 37 CFR 1.13 (6) MONTHS from the mailing date of this communication. Initiation from the mailing date of this communication. Initiation from the mailing date of this communication. In the second of the maximum statutory period we or reply within the set or extended period for reply will, by statute, by received by the Office later than three months after the mailing patent term adjustment. See 37 CFR 1.704(b).	6(a). In no event, however, may a reply be time within the statutory minimum of thirty (30) days ill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONEI	nely filed s will be considered timely. the mailing date of this communication. O (35 U.S.C. § 133).			
Status	•					
1)⊠ R	esponsive to communication(s) filed on <u>06 De</u>	ecember 2004.				
2a)⊠ Th	This action is <b>FINAL</b> . 2b) This action is non-final.					
3) <u></u> Si	S) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is					
cle	osed in accordance with the practice under E	x parte Quayle, 1935 C.D. 11, 45	3 O.G. 213.			
Disposition	of Claims					
4)⊠ Cl 4a 5)□ Cl 6)⊠ Cl 7)□ Cl	laim(s) 6-10 and 24 is/are pending in the apple of the above claim(s) is/are withdraw aim(s) is/are allowed.  aim(s) 6-10 and 24 is/are rejected.  aim(s) is/are objected to.  aim(s) are subject to restriction and/or	n from consideration.				
Application	Papers					
	e specification is objected to by the Examiner					
10)☐ The drawing(s) filed on is/are: a)☐ accepted or b)☐ objected to by the Examiner.						
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).						
	eplacement drawing sheet(s) including the correction is objected to by the Exa	-	• •			
Priority und	ler 35 U.S.C. § 119					
12) Ac a) 1. 2. 3.	knowledgment is made of a claim for foreign   All b) Some * c) None of: Certified copies of the priority documents	have been received. have been received in Application ty documents have been received (PCT Rule 17.2(a)).	on No d in this National Stage			
			•			
Attachment(s)						
	References Cited (PTO-892)  Draftsperson's Patent Drawing Review (PTO-948)	4) Interview Summary ( Paper No(s)/Mail Dat				
3) 🔲 Informati	on Disclosure Statement(s) (PTO-1449 or PTO/SB/08) o(s)/Mail Date	5) Notice of Informal Pa				



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## **DETAILED ACTION**

1. Claim 6 is amended to introduce the limitation of a convertible moiety being in a hydrophobic form. Claims 1-234 are cancelled. New claim 24 is added.

- 2. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.
- 3. Claims 6-10 are rejected under 35 U.S.C. 102(b) as being anticipated by Idogawa et al (U.S. 5,942,560).

Idogawa discloses a method of making polymer p[articles comprising admixing an aqueous carrier with an unsaturated monomer containing a hydrophobic moiety, namely methyl methacrylate (see Example 5, col.13, lines 35-44) and a suitable amount of 2-methacryloyloxyethyl succinate (monomer having convertible moiety of the instant claims, as identified by the instant specification) "ACRYL ESTER SA" and a polymerizable surfactant to form an emulsion and adding an ammonium persulfate (col.13, lines 45-52). The polymerization was continued for 3 and further 5 hours at 60C. (col.13, lines 52-56). Thus the limitations of claims 6 and 10 are fulfilled. With regard to claim 9, Idogawa teaches that in the emulsion polymerization, monomers having a reactive cross-linking group such as an epoxy group, a hydroxymethylamide group and an isocyanate group and/or multifunctional monomers having two or more vinyl groups may be blended for cross-linking. (col.col.4, lines 18-22).

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4. Claim 24 is rejected under 35 U.S.C. 103(a) as being unpatentable over Idogawa. Idogawa doews not specifically disclose the step of converting the hydrophobin moiety into a hydrophilic form. However, Idogawa suggests the Use of pH controllers if necessary. According to Idogawa, the pH controllers include, for example, ammonia, urea, monoethanolamine, diethanolamine, triethanolamine, alkaline metal salts of carbonic acid and phosphoric acid such as sodium tripolyphosphate and sodium carbonate, and hydroxides of alkaline metals such as sodium hydroxide (col.8, lines 35-42). The control and change of pH IS the very method used by Applicants to provide a conversion step as claimed in claim 24. Therefore, based on clear suggestion of Idogawa to utilize the pH controllers to modify the pH, it would have been obvious to those skilled in the art at the time the invention was made that addition of such pH controllers will result in converting of the hydrophobic moiety into a hydrophilic one.

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5. Claims 7 and 8 are rejected under 35 U.S.C. 103(a) as being unpatentable over Idogawa in view of Winnik.

With regard to claim 7 Idogawa fails to disclose that filtration is used to separate polymer particles. With regard to claim 8 Idogawa discloses water soluble basic dye vs. polymerizable dye of the instant claim.

Filtration in order to isolate the final product is a routinely used technique in the art of organic and polymer chemistry. Thus, Winnik et al (U.S. 4,795,794) discloses a method of making amphiphatic particles by dispersion polymerization process for

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affecting the preparation of particles which comprises dissolving in a suitable solvent a mixture of at least two first monomer components including at least one second monomer having covalently attached thereto a dye moiety, subsequently adding to the reaction mixture a polymerization initiator, thereafter affecting polymerization of the reaction mixture by heating, and separating the product particles therefrom (abstract). Average particle size is 0.1-20 micron (0.1 micron=100nm) (col.6, lines 40,41). The resulting mixture from polymerization, i.e. the product particles are separated from the reaction mixture by usual known techniques, including filtration (col.4, lines 34). Here the term dispersion polymerization vs. emulsion polymerization is to distinguish the resulting solid dispersion of particles, from the *initial* liquid droplets emulsion of monomers. Therefore, basically the process is the same, but by emulsion the term means the initial reaction mixture, and in dispersion, the term means the resulting polymer particles, which are obtained in either process.

Based on the similarity of the problems intended to be solved by Winnik, Idogawa and the instantly claimed invention, on the known use of filtration as a separation method (the nature of the problem to be solved) and on recognition by Winnik the equivalency of the known separation methods, including filtration, it would have been obvious to those skilled in the art at the time the invention was made to filter the resulting polymer in order to achieve better isolation of particles, and thus to arrive at the instantly claimed subject matter.

It would have been also obvious to those skilled in the art to include the polymerizable dye monomer of Winnik in lieu of mixing a dye with vinyl monomer of

Igogawa in order to achieve more homogeneous distribution of dye throughout the particle.

## Response to Arguments

6. Applicant's arguments with respect to claims 6-10 and 24 have been considered but are most in view of the new ground(s) of rejection.

## Conclusion

7. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Tatyana Zalukaeva whose telephone number is (571) 272-1115. The examiner can normally be reached on 9:00 - 5:30.

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, David Wu can be reached on (571) 272-1114. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Tatyana Zalukaeva Primary Examiner Art Unit 1713

Galuka S

February 24, 2005